SEMESTER I QP CODE 1912 REG NO.



## P.R. GOVERNMENT COLLEGE (AUTONOMOUS), KAKINADA PG ODD SEMESTEREND EXAMINATIONS-JULY-2022 M. Sc., : ORGANIC CHEMISTRY & ANALYTICAL CHEMISTRY

**PAPER 2: INORGANIC CHEMISTRY** 

DATE 13.04.2023 SESSION FN MAX. MARKS 75 TIME 3 HRS

## SECTION -A (4×15=60)

Answer all the Questions Each Question Carries 15 Marks

4×15=60 marks

1. A) Discuss the structures of trinuclear and hexanuclear metal clusters

(Or)

- B) Explain the structure and bonding in Mo<sub>2</sub> Cl<sub>8</sub>-4
- 2. A) Write the preparation, structure and bonding in Ferrocene

(Or)

- B) Explain the systhesis, structure and reactions of Dynitrogen complexes.
- 3. A) Discuss inert and labile complexes

(Or)

- B) Explain Pearson's concept of Hard and Soft acid and bases with suitable examples.
- 4. A) Discuss the mechanism of acid and base hydrolysis of Co(III)Complex

(Or)

B) Discus the mechanism of outer sphere electron transport reactions in octahedral complexes.

## **SECTION -A (5×3=15)**

Write any FIVE questions each question carries 3 marks (5×3=

(5×3=15 marks)

- Polyatomic cluster
- 6. Structure and magnetic properties of Re<sub>2</sub>(RCOO)<sub>4</sub>X<sub>2</sub>
- 7. With suitable example explain 18 electron rule
- 8. Iso electric relation ship
- 9. Elaborate Chelate effect
- 10. Inner sphere electron transfer Mechanism
- 11. Trans effect
- 12. Base hydrolysis of complexes